

GLYCOFLAVONOIDS OF ARTEMISIA TRANSILIENSIS. IV.

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From the epigeal part of Artemisia transiliensis P. in addition to O-glycosides [1, 2], the C-glycosides A, B, and C have been obtained. They were separated by chromatography of an aqueous methanolic extract on Kapron and by preparative chromatography on paper in 15% CH<sub>3</sub>COOH. The physicochemical properties of the substances studied are given in the table.

Substance	Mp, °C	[α] <sub>D</sub> <sup>20</sup> , deg	UV spectra, λ <sub>max</sub>				R <sub>f</sub>	
			in ethanol	in ethanol and Na acetate	in ethanol and Na ethoxide	in ethanol and zirconyl chloride	BAW (4:1:5)	15% CH <sub>3</sub> COOH
A	230-231	+54,5 (c 0,2; dimethyl-formamide).	236	380	395	365*	0.26	0.6
			274	282	280	348 305 284		
B	234-236	+99,9 (c 0,2; dimethyl-formamide).	234	364	400	368*	0.26	0.49
			272	282	282	346 302 282		
C	245-246 (pass)	+52 (c 0,4; methanol).	334	350	405	362*	0.5	0.66
			270	277	278	346 284		
D	342-343	-	236	378	395	382	0.94	0.1
			268	280	268	355 308 284		

\*Low-intensity absorption band.

Spectral study in the UV region showed that the substances under consideration are flavone derivatives with free OH groups in positions 5, 7, and 4'. Their positions on a chromatogram and the results of elementary analysis permit the assumptions that substances A and B are diglycosides, and substance C is a monoglycoside. These glycosides do not undergo enzymatic hydrolysis [1]. Their hydrolysis by Kiliani's method gave an apigenin C-glucoside, and traces of arabinose. On milder acid hydrolysis (5-10% solution of HCl in 50% methanol, 100° C) of the glycosides A and B, four substances were detected: two of them were identical with the glycosides under study and two had different R<sub>f</sub> values in 15% CH<sub>3</sub>COOH (0.37 and 0.12). The mixture was separated into the individual substances by preparative paper chromatography. Each compound isolated was hydrolyzed under the same conditions and gave the same four isomers. Under these conditions the glycoside C was isomerized into the second substance with R<sub>f</sub> 0.38 and 0.48. Isomerization is probably due to the rotation of the carbohydrate substituent around a C-C bond [3]. The structural features of the carbohydrate substituents were established by differential analysis in the IR region of the spectrum. The reduction in the bathochromic shifts of the zirconyl complexes of the glycosides show substitution in position 6 [4].

Thus, substance A may be characterized as apigenin 6,8-C-β,α-diglucopyranoside, which has been described in the literature [5, 6], substance B as a rotation isomer of A, and substance C as apigenin 6-β,α-glucopyranoside.

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